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Carboxylic Acids,
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Define each one ...

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= 12.00 mmol - 5.40
mmol Unreacted
ethanoic acid = 6.60
mmol Since 18.00 mL
of NaOH(aq) was

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added to 20.00 mL of ethanoic acid solution, the total volume is now 38.00 mL. Solution:

The concentration of unreacted ethanoic acid, $[\text{CH}_3\text{COOH}(\text{aq})]$, can be determined:

$$\frac{23.2 \text{ mL} \times 0.100 \text{ mol/L}}{38.00 \text{ mL}} = 0.0611 \text{ mol/L}$$

$[\text{CH}_3\text{COOH}(\text{aq})] = 6.11 \text{ mmol/L}$

Section 8.7:
Acid-Base Titration
Tutorial 1 Practice,
page 547

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2 Chemistry, First
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An organic compound consists of molecules made from carbon, hydrogen and oxygen atoms. There can be other atoms – for example, nitrogen, sulfur. 1.19 1.

Acetylsalicylic acid, the common mild pain reliever, is a derivative of the natural product, salicin, obtained from willow bark. 2.

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Chemistry 12 - Chapter 3 Quiz. True/False.

Indicate whether the sentence or statement is true or false. 1. The region in space where an electron is most likely to be found is called an energy level.

2. Electron

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configurations are often condensed by writing them using the previous noble gas core as a starting point. ...

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