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Content Review for the AP Chemistry Exam. Chapter 8. Big Idea #6: Equilibrium, Acids and Bases, Titrations, and Solubility. CHAPTER 8 ANSWERS AND EXPLANATIONS. Multiple-Choice. 1. A The equivalence point is defined as

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the point at which the moles of acid is equal to the moles of base. The moles of NaOH is equal to $(1.0 \text{ M})(0.0200 \text{ L})=0.0200 \text{ mol}$.

CHAPTER 8 ANSWERS AND EXPLANATIONS - Equilibrium, Acids

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Bonds are attractive forces that hold groups of atoms together and make them function as a unit. [Defined in 25 words or less, but leaves out LOTS of details!] Bonding relates to physical ...

AP Chemistry Chapter 8 General Bonding Test Review

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AP Chemistry Chapter 11 Review
Questions Multiple-choice exercise.
Choose the correct answer for each
question. Show all questions <= =>
How many milliliters of 6.0 M HNO₃ are
needed to prepare 500 mL of 0.50 M
HNO₃? ? 0.25 mL ? 40 mL ? 15.76 mL ?
300 mL ?

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